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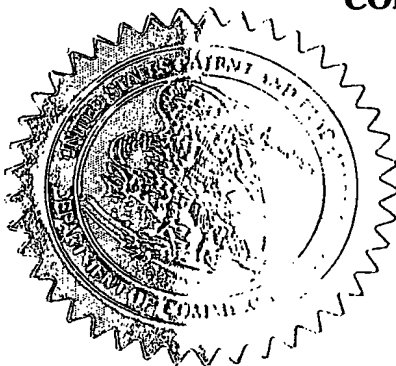
RELATED PCT APPLICATION NUMBER: PCT/US04/07247

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PROVISIONAL APPLICATION COVER SHEET

This is a request for filing a PROVISIONAL APPLICATION under 37 CFR 1.53(b)(2)

J1036 U.S. PTO

60/453231



Docket Number			P-122314.02 (PRO)	Type a plus sign (+) inside this box ->	+
INVENTOR(S)/APPLICANT(S)					
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Title of the Invention					
CONTROLLED LOW STRENGTH FLOWABLE FILL COMPOSITION					
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STATE	Texas	ZIP CODE	78205	COUNTRY	United States
ENCLOSED APPLICATION PARTS (check all that apply)					
<input checked="" type="checkbox"/> Specification/claim: Number of Pages: 10			<input type="checkbox"/> Combined Declaration/Power of Attorney		
<input checked="" type="checkbox"/> Drawing(s): Number of Sheets: 1			<input type="checkbox"/>		
METHOD OF PAYMENT (check one)					
<input checked="" type="checkbox"/> A check or money order is enclosed to cover the provisional filing fees				PROVISIONAL FILING FEE AMOUNT (\$)	\$80.00
<input checked="" type="checkbox"/> The Commissioner is hereby authorized to charge any deficiency in fees and credit Deposit Account Number: 07-2400					

☒ Applicant qualifies as a small entity under 37 CFR 1.9(c).

This invention was made by an agency of the United States Government or under contract with an agency of the United States Government.

☒ No.☐ Yes, the name of the U.S. Government agency and the Government contract number are:

Respectfully submitted,

SIGNATURE

Thomas E. Sisson, Reg. No. 29,348

Date: March 7, 2003

☐ Additional inventors are being named on separately numbered sheets attached hereto.

CERTIFICATE OF EXPRESS MAILING

I hereby certify that this paper (along with any paper referred to as being attached or enclosed) is being deposited on the date shown below with the United States Postal Service, as Express Mail Post Office to Addressee (37 CFR 1.10), Mailing Label No. EL692001553US, addressed to the Commissioner for Patents and Trademarks, Washington, D.C. 20231.

Date:

3-10-03Bianca Grossweiler
Bianca Grossweiler

1 Title: CONTROLLED LOW STRENGTH FLOWABLE FILL COMPOSITION

2 Inventors: Jerry Setliff, Scott F. Timmons, and Clinton W. Pike

3

4

BACKGROUND OF THE INVENTION

5 This invention relates to Controlled Low-Strength Mixtures (CLSMs), or
6 flowable back-fills. This class of materials has utility as pipe bedding materials where
7 they are used to both protect the pipe from external agents and internal loads. They
8 have also been used as an erosion barrier in embankments and as a mine fill material.
9 CLSMs typically have strengths of less than 2000 psi and, in cases where removal is
10 contemplated, less than 200 psi for ease of removal. The material should be initially
11 in the form of an easily pumpable, self-leveling slurry. Rapid early strength
12 development (approximately 50-70 psi) is a desirable property and is currently not
13 obtainable with commercial products without the penalty of high strength
14 development at later stages. U.S. Patent No. 5,106,422 discloses Class C Fly ash in a
15 rapid setting flowable backfill composition and method for its use.

16 However, such existing compositions are based upon the use of either Portland
17 cement or Class C fly ash used individually or in combination as the hydraulic cement
18 component of the CLSM system. Typically these cementitious materials are used at
19 less than 5% by weight in the case of Portland cement or as much as 50% in the case
20 of Class C fly ash with the remainder being some form of aggregate, usually fine sand
21 or soil from the spoil with small amounts of additional rock and gravel or Class F fly
22 ash. Cement-based materials can take days to hydrate, cure, and achieve even a

1 modest strength of 50 psi which is typically the minimum strength required for a man
2 to walk upon the surface of the bedding material and represents the minimum safe
3 time before the cover fill may be placed. Class C fly ash based systems may take as
4 long as four hours to hydrate, cure, and achieve this strength. In many cases, locally
5 available Class C fly ash is not desirable for use in these types of product due to slow
6 hydration, cure, and set times and low strengths. Strength may be compensated for by
7 the use of additional Class C fly ash but the cost of the additional fly ash may result in
8 cost prohibitive products.

9 Thus, the system of the present invention minimizes the down time before
10 cover fill may be placed and represents a significant savings of both time and money
11 for the user. Furthermore, the present inventive composition and method allow for
12 control variability in strength and hydration, cure, and set times of a CLSM system
13 utilizing Class C fly which results in a Class C-based flowable fill capable of
14 competing in markets previously inaccessible.

15 BRIEF DESCRIPTION OF THE DRAWINGS

16 Fig. 1 is a graphic representation of the effect of set time of Class C mortars
17 with lime.

18 DETAILED DESCRIPTION OF THE PREFERRED EMBODIMENT

19 Class C fly ash as defined in ASTM C 618 is a coal combustion product that
20 meets particular size requirements and mineralogical specifications. A typical
21 chemical composition for this class of fly ash is as follows:
22

		<u>Percent by Weight</u>
1		
2	Silicon dioxide (SiO_2) plus aluminum	
3	oxide (Al_2O_3) plus iron oxide (Fe_2O_3), min.	50.0
4	Sulfur trioxide (SO_3), max.	5.0
5	Moisture content, max.	3.0
6	Loss on ignition, max.	6.0

7
8 This is a rather broad description for this class of material, and significant
9 variability may exist for materials conforming to this requirement. The variability
10 manifests itself as differences in hydration and set time and strength between several
11 samples of Class C fly ash either from the same or different sources. It has been found
12 that a major factor contributing to variability is the amount of available calcium
13 present in the sample. Additionally, soluble iron content contributes to slow setting
14 times. Furthermore, the addition of small amounts of calcium to Class C fly ash has
15 no deleterious effects upon flowable fill and can accelerate the rate of hydration and
16 cure while minimizing the differences in set time and strength of flowable fill mixtures
17 containing Class C fly ash.

18 Where soluble iron is present in sufficient quantity and extra calcium alone is
19 inadequate to accelerate the rate of hydration, iron chelating compounds may be
20 added, even in very small amounts, to offset the soluble iron effect. The iron chelating
21 compound may include:

22 sodium chloride, sodium thiosulfate, triethanolamine, diethanolamine,
23 polyethyleneimine, amino-substituted acrylic monomers or polymers,
24 morpholine and substituted morpholine compounds, urea, guanidine salts,
25 pyrole and pyrole compounds, polyvinylpyrole, imidazole compounds,
26 pyrazoles, pyridine and pyridine compounds (especially ortho alkoxy-

1 substituted pyridines), amino phenol (especially ortho amino phenol), amino
2 cresol, ortho anisidine, amine acetate surfactants (such as Armac HT and
3 Armac 18D-40 from Akzo Nobel Chemicals), amine oxide surfactants (such as
4 Ammonyx series of surfactants from Stepan Company, Schercamox series of
5 surfactants from Scher Chemicals, Foamox series of surfactants from Alzo,
6 Inc., Chemoxide series of surfactants from Chemron Corp.), amine surfactants
7 (such as the Armeen and Redicote series of surfactants from Akzo Nobel
8 Chemicals, the Incromine series of surfactants from Croda, Inc., the Tealan
9 series of surfactants from R.I.T.A. Corp.), and mercapto surfactants (such as
10 Burco TME from Burlington Chemicals).

11 The iron chelating compound may be in quantities in the range of 0.01% or
12 5.0% by weight. Effective results have been obtained and reasonably should be
13 obtained from chelting agents or compounds selected from the group consisting of an
14 alkanolamine, a polymer of ethyleneimine, a block copolymer containing
15 polyethyleneimine segments, an amino-substituted polymer of acrylic acid, the salt of
16 an amino-substituted polymer of acrylic acid, a carboxyated amine compound, a salt
17 of a carboxyated amine compound, ethylenediaminetetraacetic acid and salts thereof;
18 nitrilotriacetic acid and salts thereof, an amine substituted surfactant, an amine oxide
19 substituted surfactant, and a guanidine salt.

20 The following examples illustrate the nature of the present invention. Set
21 times were determined when a 0.25" diameter penetrometer needle provided a reading
22 of 200 psi on insertion to a depth of 1.0".

1

2 Example 1

3 Coal Fly ash from Deeley Power Plant, San Antonio, Texas, as obtained and
4 used as received. 50 grams of Class C Fly ash, 250 grams ASTM C 33 graded washed
5 silica sand (Espey Sand, San Antonio, Texas) and 35 mL deionized water were mixed
6 for 1 minute and poured into a 2" cube mold. The set time was determined to be 62
7 minutes as shown in Table 1 below.

8 Examples 2-8

9 Coal Fly ash from Deeley Power Plant, San Antonio, Texas, was obtained and
10 used as received. 50 grams of Class C Fly ash, 250 grams ASTM C 33 graded washed
11 silica sand (Espey Sand, San Antonio, Texas), varying amounts of type S hydrated
12 lime and 35 mL deionized water were mixed for 1 minute and poured into a 2" cube
13 mold. The set time for these examples are shown in Table 1 for the varying amounts
14 of lime.

15 Examples 9-13

16 Coal Fly ash from Scherer Power Plant, Atlanta, Georgia, was obtained and
17 used as received. 50 grams of Class C Fly ash, 250 grams ASTM C 33 graded washed
18 silica sand (Espey Sand, San Antonio, Texas), varying amounts of type S hydrated
19 lime and 35 mL deionized water were mixed for 1 minute and poured into a 2" cube
20 mold. The set times for these examples are shown in Table 1 for the varying amounts
21 of lime.

1 A graphic representation of the effect of set time of these Class C motars with
 2 lime of varying amounts is shown in Fig. 1.

3 Table 1. Set times of Class C Fly ash motar cubes containing varying amounts
 4 of type S lime.

Lime (grams)	Example Number	Set Time (minutes) Deeley	Example Number	Set Time (minutes) Scherer
0.00	1	62	9	348
0.07	2	51	-	-
0.13	3	26	-	-
0.25	4	12	10	303
0.50	5	10	11	71
0.75	6	8	12	37
1.00	7	12	13	76
1.50	8	9		

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Example 14

Ingredient	Amount (g)
Concrete Sand	0
Type C Fly Ash	100
Hydrated Lime	.003
Triethanolamine	.04
Water	25

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The dry ingredients were mixed together and the water was added with mixing until a smooth, pourable consistency was obtained. The set time was 17 minutes.

Example 15

Ingredient	Amount (g)
Concrete Sand	200
Type C Fly Ash	100
Hydrated Lime	.3
Triethanolamine	.48
Water	39

17

1
2 The dry ingredients were mixed together and the water was added with mixing
3 until a pourable consistency was obtained. The set time was 17 minutes.
4
5

6 Example 16
7

Ingredient	Amount (g)
Concrete Sand	250
Type C Fly Ash	50
Hydrated Lime	7.5
Triethanolamine	.1
Water	40

8
9 The dry ingredients were mixed together briefly and the water and
10 triethanolamine added with continued mixing. The set time was 23 minutes.
11
12
13

1
2 CLAIMS:
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- 5 1. A rapid setting, controlled low strength composition of Class C fly ash
6 comprising hydrated lime in the amount of 0.1% to 15% by weight and an iron
7 chelating compound in the amount of from 0.01% to 5% by weight sufficient to
8 accelerate the hydration and set time of said fly ash.
- 9 2. A rapid setting, controlled low strength composition of Class C fly ash
10 comprising hydrated lime in the amount of 0.1% to 15% by weight of fly ash and an
11 iron chelating compound in the amount of from 0.01% to 5% by weight sufficient to
12 accelerate the hydration and set time of said fly ash, and a filler material in the amount
13 of 1:10 to 10:1 parts by weight.
- 14 3. A method by which the hydration and set time of a cementitious mixture
15 containing Class C fly ash is accelerated comprising the step of adding hydrated lime
16 in the amount of 0.1% to 15% by weight of and an iron chelating compound in the
17 amount of from 0.01% to 5% by weight cementitious material to said cementitious
18 mixture.
- 19 4. A rapid setting, controlled low strength composition of Class C fly ash
20 comprising a calcium source in the amount of 0.1% to 15% by weight and an iron
21 chelating compound in the amount of from 0.01% to 5% by weight sufficient to
22 accelerate the hydration and set time of said fly ash.
- 23 5. The composition of claim 4 wherein said source is quicklime.

1 6. The composition of claim 4 wherein said calcium source is selected from the
2 group consisting of calcium nitrate, calcium nitrite, calcium formate, calcium acetate,
3 calcium propionate, calcium lignosulfonate, calcium oxide, calcium hydroxide,
4 calcium hypochlorite, anhydrous calcium sulfate, calcium sulfate dihydrate, and
5 calcium sulfate hemihydrate.

6 7. The composition of claim 4 wherein said calcium source is a circulating
7 fluidized bed coal ash containing free lime in the amount of 0.25% to 70% by weight
8 of Class C fly ash.

9 8. The composition of claim 2 wherein said filler material is selected from the
10 group consisting of Class F fly ash, silica sand, dolomitic calcium carbonate sand,
11 limestone sand, expanded perlite, expanded styrofoam, bottom ash, slag, foundry sand,
12 expanded shale, clay, ground granite sand, pumice and gravel.

13 9. The composition of claim 4 wherein said iron chelating compound is selected
14 from the group consisting of an alkanolamine, a polymer of ethyleneimine, a block
15 copolymer containing polyethyleneimine segments, an amino-substituted polymer of
16 acrylic acid, the salt of an amino-substituted polymer of acrylic acid, a carboxyated
17 amine compound, a salt of a carboxyated amine compound, ethylenediaminetetraacetic
18 acid and salts thereof; nitrilotriacetic acid and salts thereof, an amine substituted
19 surfactant, an amine oxide substituted surfactant, and a guanidine salt.

20

ABSTRACT OF THE DISCLOSURE

1
2 A rapid setting, controlled low strength composition of Class C fly ash is
3 provided having a quantity of hydrated lime and an iron chelating compound in an
4 amount sufficient to accelerate the hydration and set time of the fly ash. In some
5 examples, a filler material is added. A method for acceleration of the hydration and
6 set time of a cementitious mixture is provided wherein hydrated lime is added to the
7 cementitious mixture in an amount in the range of 0.1% to 15% by weight and an iron
8 chelating compound in an amount in the range of 0.01% to 5.0% by weight of the
9 cementitious material. Further, a calcium source and an iron chelating compound may
10 be added to a Class C fly ash to accelerate the hydration and set time of the ash.

Effect of Set Time of Class C Mortars with Lime

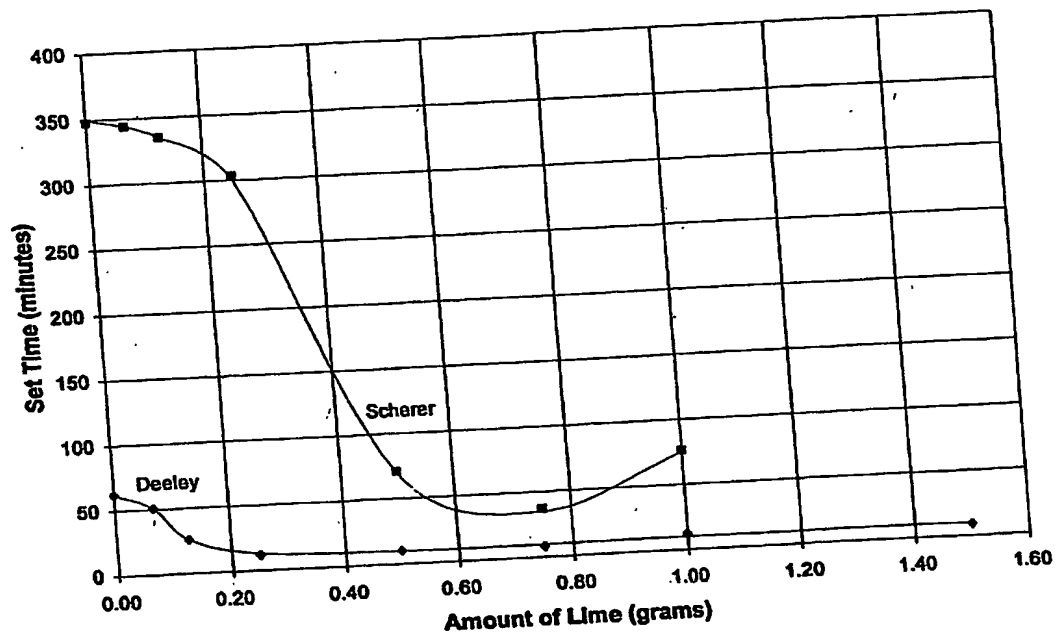


FIG 1.